(essential review

Using the words provided, fill in the blanks below to form a thorough review of Chapter 7, "The Basics of Chemistry and Electricity." Words or terms may be used more than once or not at all.

acid	combustion	hydrophilic	oxidizing
\ alcohol	compound	infrared	physical
alkaline	conductor	inorganic	polarity
alkanolamines	converter	light rays	radiant energy
alternating	density	lipophilic	rectifier
/ ampere	direct	liquids	solute
anaphoresis	disincrustation	matter	solvents
anode	electricity	miscible	surfactant
atom	electrode	mixtures	Suspension
blue	element	molecule	therapeutic
cathode	emulsions	ointments	vaporizer
chemical	formaldehyde	organic	volatile
chemistry	galvanic	overheating	water
circuit breaker	high-frequency	oxidation	

1.	A solution that has a pH less than 7 has an pH, and a solution that
	has a pH higher than 7 has an pH.
2.	A change is a change in the form of a substance, without the
	formation of a new substance. A change occurs when a new
	substance is formed.
3.	A is a substance that acts as a bridge to allow oil and water to mix, or
	emulsify.
4.	is a readily evaporating, colorless liquid obtained by the fermentation
	of starch, sugar, and other carbohydrates.
5.	An is the smallest particle of an element that is capable of showing
	the properties of that element.
6.	Anything that occupies space is defined as
7.	Characteristics of physical properties include, specific gravity,
	hardness, odor, and color.
8.	are formed when two or more immiscible substances, such as oil and
	water, are united with the aid of a binder.

essential review continued

_	chemistry is the branch of chemistry that deals with all substances
	the state of the second
	, and gasser
4 4	pastes, pomades, and styling water
– v 12.	rith any combination of petrolatum, oil, and wax. chemistry is the branch of chemistry that deals with all substances in
٧	which carbon is present. agents are substances that readily release oxygen.
14. 5	Solvents are classified as and immissions.
16. 5	Surfactant molecules have two ends: and A is a state in which solid particles are distributed throughout a liquid
1	medium.
18.	The universal solvent is
	The basic unit of all matter is an The science that deals with the composition, structure, and properties of matter is
	To measure the pH of products, use paper.
22. 23.	When a substance is made up of two or more different significant.
24. 25.	When elements combine physically, they are called When oxygen combines with other substances so rapidly that light energy as well as heat is created, the process is known as
26.	is a form of energy that produces magnetic, offerment,
	effects.
27.	A is a substance that permits electrical can be applied to the A steamer or produces moist, uniform heat that can be applied to the
	head or face. A is used to change direct current into alternating current, and a is used to change direct current to direct current.
30	A positive electrode is called a/an, and a negative
	a/an

essential review continued

31.	A fuse is a safety device that prevents theof electric wires.
32.	An amp or is the unit that measures the strength of an electric current
	(the number of electrons flowing through a wire).
33.	An is an applicator that directs the electric current from the machine
	to the client's skin.
34.	is the process of forcing liquids into the tissues from the negative
	toward the positive pole.
35.	Artificial light rays are produced by using an electrical apparatus called a
	lamp.
36.	current is a constant, even-flowing current, traveling in one direction,
	while current is a rapid and interrupted current, flowing first in one
	direction, then in the opposite.
37.	Do not use the negative current on skin with broken capillaries or
	pustular acne, or on a client with high blood pressure.
38.	In modern electric wiring, the fuse has largely been replaced by the
39.	rays make up 60 percent of natural sunlight.
40.	Therapeutic lamps are used to produce artificial in the salon.
	The negative or positive state of electric current is
	The current is characterized by a high rate of oscillation or vibration.
43.	The process used to soften and liquify grease deposits in the hair follicles is
44.	The light contains few heat rays and has some germicidal and
	chemical benefits.
45.	creates a chemical change in the identity of a substance.
	The dissolved substance in a solution is known as a
47.	alcohols are those that evaporate easily.
48.	Substances that are used to neutralize acids or raise the pH of many hair products are
40	known as
+ ∀.	A preservative used in cosmetics that is toxic to inhale, a strong irritant, and a
50	Another name for electromegnetic registion in
uu.	AUGUETOR DOMO for alactromagnatic radiation is

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Word Search

After determining the correct words from the clues provided, locate the words in the word search puzzle.

	Solution having a pH below 7.0
	Readily evaporating, colorless liquid
	Solution having a pH above 7.0
	Colorless gas with pungent odor, composed of hydrogen and nitrogen
P-3	The smallest particle of an element that still retains the properties of that element
tra	Science that deals with the composition, structures, and properties of matter
	Rapid oxidation of any substance
	Chemical combination of two or more atoms of different elements
	The simplest form of matter
	Mixture of two or more immiscible substances united with the aid of a binder or
	emulsifier
	Sweet, colorless, oily substance formed by the decomposition of oils, fats, or
•	fatty acids
	Capable of combining with or attracting water
	Not capable of being mixed
	Capable of combining with or attracting oil
	Any substance that occupies space, has physical and chemical properties, and
	exists in the form of a solid, liquid, or gas
	Two or more atoms joined chemically
	Chemical reaction that combines an element or compound with oxygen to
	produce an oxide
	Blended mixture of two or more solids, liquids, or gaseous substances
	Surface active agent
	State in which solid particles are distributed throughout a liquid medium

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Word Scramble

Scramble	Correct Word(s)
eotuls	
	Clue: Any substance that dissolves into a liquid and forms a
	solution
etawr	
	Clue: A universal solvent
iicilbesmm	
	Clue: Not mixable
lelitavo	
	Clue: Easily evaporated '
Inmioseu	
	Clue: Formed when two or more immiscible substances are
	united with the aid of a binder
melsbiic	
	Clue: Mixable
peword	
	Clue: Physical mixture of two solids
setpsa	
	Clue: Semisolid mixture made of petrolatum, oil, and wax
snoituol	
	Clue: Evenly dispersed mixtures of two or more kinds of molecules
tnevols	
	Clue: Any substance that is able to dissolve another substance
iosnnpesus	
	Clue: Mixtures of one type of matter in another type of matter
nntsiemto	
	Clue: Semisolid mixtures of organic substances and a
	medicinal agent

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Conductors and Types of Electricity

	ace provided.	ctors or nonconductors of electri	
·		Silk	Asbestos
			Tar
			Water
			Water
	Human body	Carbon	Linen
	Cement	Acid or salt solutions	
			and an alternating curre
n your c	own words, explain the diffe		and an alternating curre
n your c	own words, explain the diffeurrent:	erence between a direct current a	and an alternating curre
n your c	own words, explain the diffeurrent:	erence between a direct current a	and an alternating curre
n your c	own words, explain the diffeurrent:ing current:	erence between a direct current a	and an alternating curre

Matching Exercise—Electrical Measurements

Match each of the following essential terms with its definition.

 Volt
 Amp
Milliampere
 Ohm
 Watt
 Kilowatt

- **1.** Measurement of how much electric energy is being used in one second.
- 2. 1/1,000 of an ampere. The current for facial and scalp treatments is measured in this manner.
- **3.** The unit of measurement for the strength of an electric current (the number of electrons flowing through a wire).
- **4.** Unit for measuring the pressure that forces the electric current forward.
- 5. The electricity in your house is measured with this unit.
- **6.** This unit measures the resistance of an electric current. Unless the force is stronger than the resistance, current will not flow through the wire.

Safety of Electrical Equipment

Fill in the blanks for the safety precautions that should be followed to avoid accidents and ensure greater client satisfaction.

1.	All the electrical appliances you use should be
2.	Read all before using any electrical equipment.
3.	all appliances when not in use.
4.	all electrical equipment regularly.
5.	Keep all wires, plugs, and equipment in good
6.	Use only one plug to each
7.	You and your client should avoid contact with and metal surfaces when using electricity.
8.	Do not leave your client unattended while to an electrical device.
9.	Keep electrical cords off the and away from people's feet.
10.	Do not attempt to around electric outlets while equipment is plugged in.
11.	Do not touch two objects at the same time if either is connected to an electric current.
12.	Do not step on or place on electrical cords.
13.	Do not allow electrical cord to become as it can cause a short circuit.
14.	Disconnect appliances by pulling on the, not the cord.
15.	Do not attempt to electrical appliances unless you are qualified.

			Annual Committee
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Matching Exercise

Match the following essential terms with their identifying phrases or definition.

 Alcohol
 Acid
 Ammonia water
 Disincrustation
 Atom
 Glycerine
 Petrolatum
 Lipophilic
 Redox
 Silicone

- **1.** Special type of oil used in hair conditioners and as a water-resistant lubricant for the skin.
- 2. A sweet, colorless, odorless, oily substance formed by the decomposition of oils, fats, or fatty acids.
- **3.** Process used to soften and emulsify grease deposits and blackheads in the hair follicles.
- **4.** Chemical reaction in which the oxidizing agent is reduced and the reducing agent is oxidized.
- 5. Having an affinity or attraction to fat and oils.
- 6. Another term for petroleum jelly.
- 7. The smallest particle of an element that still retains the properties of that element.
- **8.** A colorless liquid with a pungent odor, composed of hydrogen and nitrogen in a water solution.
- **9.** A colorless liquid obtained by the fermentation of starch, sugar, and other carbohydrates.
- **10.** Having a pH below 7.0.

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